



QUANTUM MECHANICS

UNIT II Quantum Mechanics Lecture-3



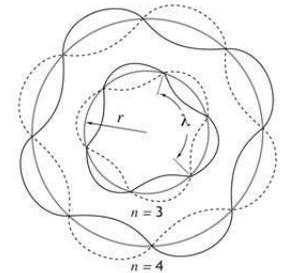
MODERN PHYSICS • XXIII.iii • Wave Mechanics and Atomic Theory

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The De Broglie Wavelength

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

- λ = wavelength
- h = Planck's constant ($6.63 \times 10^{-34} \text{ J} \cdot \text{s}$)
- p = momentum
- m = mass
- v = speed



De Broglie's extension of the concept of particle-wave duality from photons to include all forms of matter allowed the interpretation of electrons in the Bohr model as standing electron waves. De Broglie's work marked the start of the development of wave mechanics.

$$E\Psi = -\frac{\hbar^2}{2m} \frac{d^2\Psi}{dx^2} + V\Psi$$





WAVE VELOCITY OR PHASE VELOCITY

- The velocity of advancement of a monochromatic wave (i.e., a wave of single frequency and wavelength) in a medium is known as *wave velocity, or phase velocity*.

$$v_p = \frac{\omega}{k}$$

- The velocity of propagation of planes of constant phase through a medium is known as *wave velocity, or phase velocity*.

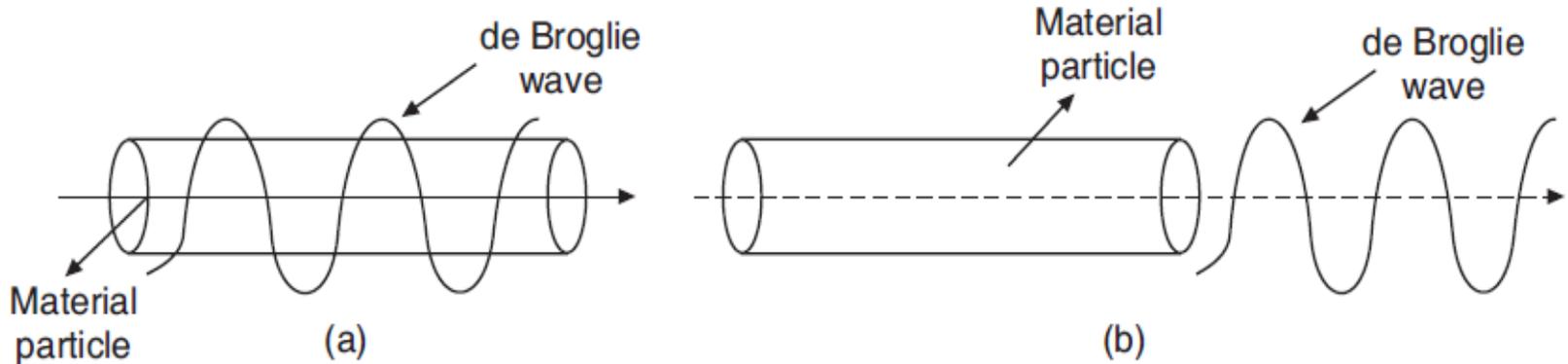


NEED FOR WAVE PACKET REPRESENTATION

- For the material particle (including electron and proton), u is always less than c .
- It means that, according to $v=c^2/u$, the phase velocity of the wave associated with the material particle is always greater than c .
- Now, it can be concluded that the particle and its corresponding de Broglie wave cannot travel together.
- Hence, the particle should be left behind to its de Broglie wave.



NEED FOR WAVE PACKET REPRESENTATION

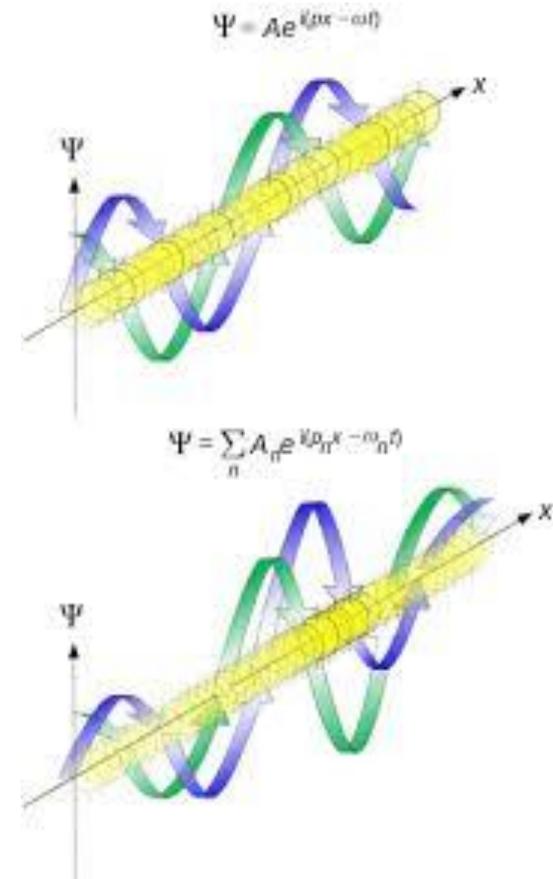


(a) Material particle and de Broglie wave are moving together and **(b)** material particle is left behind because $v_{\text{particle}} < c$ and $v_p > c$



NEED FOR WAVE PACKET REPRESENTATION

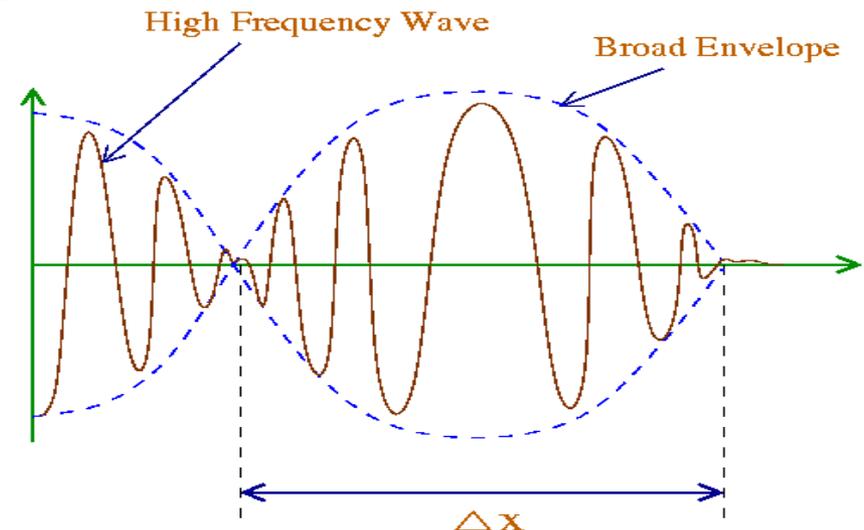
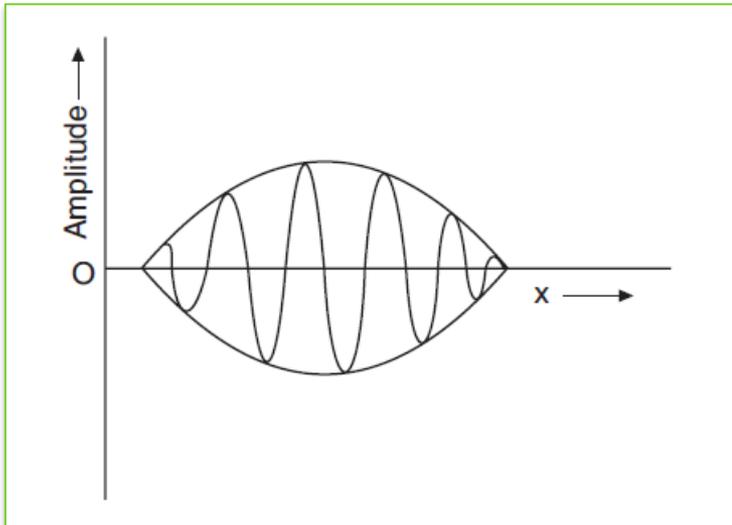
- From the earlier discussion, it seems that the particle will not be able to keep pace with the associated de Broglie wave.
- However, it is not so. In order to explain the conflicting idea of velocity relationship between material particle and its de Broglie wave, Schrödinger himself introduced the idea of wave packet.
- According to the idea of wave packet, a moving material particle is equivalent to a wave packet (number of waves) instead of a single wave.





Wave Packet

- A wave packet is the resultant of a group of waves, slightly differing in velocity and wavelength, with such phase and amplitude that they interfere constructively over a small region of space where the particle can be located.
- Outside this space, they interfere destructively so that the amplitude reduces to zero.

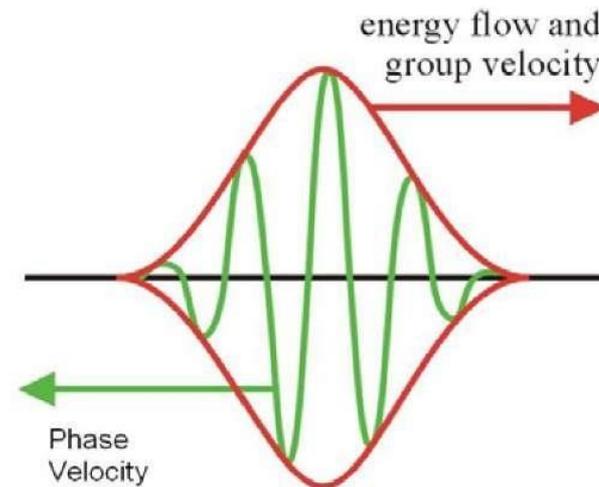




GROUP VELOCITY

The velocity with which a wave packet (or the group of waves) associated with the moving particle travels is called the *group velocity*, whereas the velocity with which the individual waves comprising the wave packet travel is referred to as *wave velocity*, or *phase velocity*.

$$v_g = \frac{d\omega}{dk}$$





EXPRESSION FOR GROUP VELOCITY

$$y_1 = a \sin (\omega_1 t - k_1 x)$$

$$y_2 = a \sin (\omega_2 t - k_2 x)$$

From the principle of superposition, the resultant displacement of these waves can be given as

$$y = y_1 + y_2$$

$$= a \sin (\omega_1 t - k_1 x) + a \sin (\omega_2 t - k_2 x)$$

$$= 2a \cos \left[\left\{ \frac{\omega_1 - \omega_2}{2} \right\} t - \left\{ \frac{k_1 - k_2}{2} \right\} x \right] \times \sin \left[\left\{ \frac{\omega_1 + \omega_2}{2} \right\} t - \left\{ \frac{k_1 + k_2}{2} \right\} x \right]$$

$$\left(\text{because } \sin A + \sin B = 2 \sin \left(\frac{A+B}{2} \right) \cdot \cos \left(\frac{A-B}{2} \right) \right)$$



EXPRESSION FOR GROUP VELOCITY

If ω_1 and ω_2 are very close to each other, then $(\omega_1 + \omega_2) / 2 = \omega$ and $(k_1 + k_2) / 2 = k$ (say). Hence, the above equation may reduce to the following form:

$$y = 2a \cos \left[\left(\frac{d\omega}{2} \right) t - \left(\frac{dk}{2} \right) x \right] \sin [\omega t - kx]$$

$$A = 2a \cos \frac{d\omega}{2} \left[t - \frac{dk}{d\omega} x \right]$$

$$A = 2a \cos \frac{d\omega}{2} \left[t - \frac{x}{v_g} \right]$$

where v_g is the group velocity. This is given as

$$v_g = \frac{d\omega}{dk}$$

or

$$v_g = \frac{\omega_1 - \omega_2}{k_1 - k_2}$$



RELATION BETWEEN GROUP VELOCITY AND WAVE VELOCITY

From the definition of phase velocity, we know that

$$v_p = \frac{\omega}{k}$$

or

$$\omega = kv_p$$

Differentiating the above equation, we get

$$d\omega = kv_p + v_p dk$$

or

$$\frac{d\omega}{dk} = v_p + k \frac{dv_p}{dk}$$

$$v_g = \frac{d\omega}{dk}$$

$$= v_p + k \frac{dv_p}{dk}$$

or

$$v_g = v_p + k \frac{dv_p}{d\lambda} \cdot \frac{d\lambda}{dk}$$

Since $k = 2\pi / \lambda$, hence, $d\lambda / dk = -2\pi / k^2$.

$$v_g = v_p - \lambda \frac{dv_p}{d\lambda}$$



RELATION BETWEEN GROUP VELOCITY AND PARTICLE VELOCITY

$$v_g = v_p - \lambda \frac{dv_p}{d\lambda}$$

$$= \lambda^2 \left[\frac{v_p}{\lambda^2} - \frac{1}{\lambda} \frac{dv_p}{d\lambda} \right]$$

$$= -\lambda^2 \frac{d}{d\lambda} \left(\frac{v_p}{\lambda} \right)$$

$$= -\lambda^2 \frac{d}{d\lambda} \left(\frac{1}{\lambda} \right)$$

$$\frac{1}{v_g} = -\frac{1}{\lambda^2} \frac{d\lambda}{dv}$$

$$\frac{1}{v_g} = \frac{d}{d\lambda} \left(\frac{1}{\lambda} \right)$$

$$\frac{1}{v_g} = \frac{1}{h} \frac{d}{dv} [2m(E - V)]^{1/2}$$

$$= \frac{1}{h} \frac{1}{dv} [2m(hv - V)]^{1/2}$$

$$= \frac{1}{h} \cdot \frac{1}{2} [2m(hv - V)]^{-1/2} \cdot 2mh$$

Group velocity = Particle velocity

$$\frac{m}{(E - V)^{1/2}}$$

$$= \left[\frac{m}{\{2(E - V)\}} \right]^{1/2}$$

$$= \frac{1}{v_{\text{particle}}}$$

$$v_g = v_{\text{particle}}$$



GROUP VELOCITY OF de BROGLIE WAVES

According to the relativistic consideration

$$m = \frac{m_0}{\sqrt{1 - v^2 / c^2}}$$

We know that

$$\omega = 2\pi\nu = \frac{2\pi E}{h} = \frac{2\pi mc^2}{h}$$

and

$$k = \frac{2\pi}{\lambda} = \frac{2\pi m v}{h} \quad \left[\because \lambda = \frac{h}{m v} \right]$$

$$\omega = \frac{2\pi m_0 c^2}{h \sqrt{1 - v^2 / c^2}}$$

and

$$k = \frac{2\pi m_0 v}{h \sqrt{1 - v^2 / c^2}}$$



GROUP VELOCITY OF de BROGLIE WAVES

Differentiating ω and k with respect to v

$$\frac{d\omega}{dv} = \frac{2\pi m_0 v}{h[1 - v^2 / c^2]^{3/2}}$$

$$\frac{dk}{dv} = \frac{2\pi m_0}{h[1 - v^2 / c^2]^{3/2}}$$

$$\begin{aligned} v_g &= \frac{d\omega}{dk} = \frac{d\omega / dv}{dk / dv} \\ &= 2\pi m_0 v / 2\pi m_0 \\ v_g &= v \end{aligned}$$



The phase velocity of sea-water waves is given by $\sqrt{(g\lambda / 2\pi)}$. Calculate the group velocity of these waves. The wavelength is 680 m. Take $g = 9.8 \text{ m/s}^2$.

Solution

The relation between group velocity (v_g) and phase velocity v_p can be given as

$$dv_p = \frac{1}{2} \sqrt{\left(\frac{g\lambda}{2\pi}\right)} \frac{1}{\lambda}$$

Given

$$= \frac{v_p}{2\lambda}$$

or

Now, by putting the value of $dv_p/d\lambda$ in Eq. (1), we get

$$\begin{aligned} v_g &= v_p - \frac{v_p}{2\lambda} \\ &= \frac{v_p}{2} \\ &= \frac{1}{2} \left[\frac{g\lambda}{2\pi} \right]^{1/2} \\ &= \frac{1}{2} \left[\frac{9.8 \times 680}{2 \times 3.14} \right]^{1/2} \\ &= 16.29 \text{ m/s} \end{aligned}$$



Example

Show that the de Broglie wave velocity is a function of wavelength even in free space.

Solution

The de Broglie wavelength of a wave associated with a particle of mass m moving with velocity v is given by

$$\begin{aligned}\lambda &= \frac{h}{mv} \\ &= \frac{h}{m_0c \sqrt{\frac{v_p^2}{c^2} - 1}} \\ &= \frac{h}{m_0c \left[\frac{v_p^2}{c^2} - 1 \right]^{1/2}}\end{aligned}$$

or

$$\left(\frac{\lambda m_0 c}{h} \right)^2 = \frac{v_p^2}{c^2} - 1$$

$$\frac{v_p^2}{c^2} = 1 + \left(\frac{\lambda m_0 c}{h} \right)^2$$

or

$$v_p = c \sqrt{1 + \left(\frac{m_0 c \lambda}{h} \right)^2}$$

It is clear from the above expression that for a particle of mass m , the wave velocity is always greater than c and it is a function of λ in free space.



Distinguish between the group velocity (v_g) and the phase velocity (v_p) of a wave packet and show that $v_p v_g = c^2$.

Solution

According to the de Broglie hypothesis, the wavelength of a wave associated with a particle of mass m moving with velocity v is given as

$$\lambda = \frac{h}{mv} = \frac{h}{p}$$

The propagation constant k is defined as

$$k = \frac{2\pi}{\lambda} = \frac{2\pi mv}{h} \tag{1}$$

If a particle has its energy E and the frequency of a wave associated with it is ν , then we can write

$$E = h\nu \quad \text{and} \quad \omega = 2\pi\nu = \frac{2\pi E}{h}$$

According to the mass energy equivalence relation, we know that

$$E = mc^2$$

Hence,

$$\omega = \frac{2\pi mc^2}{h} \tag{2}$$

We know that

$$v_p = \frac{\omega}{k} = \frac{2\pi mc^2 / h}{2\pi mv / h}$$
$$v_p = \frac{c^2}{v} \tag{3}$$

The particle is moving with a velocity v . Since the particle velocity is equal to the group velocity, i.e., $v_g = v_{\text{particle}}$, so $v = v_g$.

Now, putting this value in Eq. (3), we get

$$v_p = \frac{c^2}{v_g}$$

or $v_p v_g = c^2$



Assignment Based on this Lecture

- Define phase velocity and obtain the expression for phase velocity.
- Discuss the need for wave packet representation.
- Define group velocity and obtain the expression for group velocity.
- Obtain the relation between phase velocity and group velocity
- Prove that the particle always travel with group velocity.